## Short Communications

Contributions intended for publication under this heading should be expressly so marked; they should not excced about 1000 words; they should be forwarded in the usual way to the appropriate Co-editor; they will be published as speedily as possible. Publication will be quicker if the contributions are without illustrations.

Acta Cryst. (1967). 22, 918
The space groups and unit-cell dimensions of 2,4,6-tribromoaniline and 2,4,6-trichloroaniline. By E. O. Schlemper* and Judith Konnert, Department of Chemistry, University of Minnesota, Minneapolis, Minnesota 55455, U.S.A.
(Received 16 January 1967)

Crystals of $2,4,6$-tribromoaniline are orthorhombic, space group $P 2_{1} 2_{1} 2_{1}$, with $a=13 \cdot 46, b=14 \cdot 69, c=$ $4 \cdot 27 \AA, Z=4$. Crystals of $2,4,6$-trichloroaniline are monoclinic, space group either $P 2_{1} / m$ or $P 2_{1}$, with $a=15 \cdot 88, b=3 \cdot 86, c=13 \cdot 38 \AA, \beta=111^{\circ} 45^{\prime}, Z=4$.

In connection with some X-ray work being done on 2,4,6tribromobenzonitrile and 2,4,6-trichlorobenzonitrile, the space groups and unit cells of 2,4,6-tribromoaniline and 2,4,6-trichloroaniline were determined.

From oscillation, Weissenberg, and precession photographs (Mo $K \alpha, \lambda=0.7107 \AA$ ), crystals of 2,4,6-tribromoaniline were found to be orthorhombic with cell dimensions $a=13.462 \pm 0.008, \quad b=14.692 \pm 0.014, \quad c=4.266 \pm 0.005 \AA$. The systematic extinctions, ( $h 00, h=2 n+1 ; 0 k 0, k=2 n+1$; $00 l, l=2 n+1$ ), indicate that the space group is $P 2_{1} 2_{1} 2_{1}$. If $Z=4$, the calculated density is $2.596 \mathrm{~g} . \mathrm{cm}^{-3}$; the experimental density is $2.578 \mathrm{g.cm}{ }^{-3}$ (Jaeger, 1907).

[^0]Needle-like crystals of 2,4,6-trichloroaniline were mounted in capillaries to prevent sublimation. Precession photographs taken with Mo $K \alpha$ radiation ( $\lambda=0.7107 \AA$ ) indicate that the cell is monoclinic with the unique axis as the needle axis. A unit cell with the dimensions $a=15.875 \pm 0.006$, $b=3.863 \pm 0.005, c=13.381 \pm 0.005 \AA, \beta=111^{\circ} 45^{\prime}$ was chosen. A rough experimental density of $1.6 \mathrm{~g} . \mathrm{cm}^{-3}$ was obtained; for $Z=4$, the calculated density is $1.712 \mathrm{~g} . \mathrm{cm}^{-3}$. The systematic extinctions, $(0 k 0, k=2 n+1)$, indicate that the space group is either $P 2_{1} / m$ or $P 2_{1}$. No further X-ray work on these compounds is planned.

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## References

JaEger, F. M. (1S07). Z. Kristallogr. 42, 236.

## Acta Cryst. (1967). 22, 918

Absorption correction in the Weissenberg methods. By A.Santoro and M. Zocchi, Institute for Materials Research,
National Bureau of Standards, Washington, D.C., U.S.A.
(Received 3 November 1966)
A procedure is described for calculating the direction cosines of the incident and diffracted beams in a reference system attached to the crystal for the general case of the Weissenberg method.

Methods have been described for the calculation of the absorption correction in the equi-inclination and normalbeam techniques (Wells, 1960; Coppens, Leiserowitz \& Rabinovich, 1965; Wuensch \& Prewitt, 1965). However in some cases (Santoro \& Zocchi, 1966) it is preferable to use the Weissenberg method with arbitrary values of the angle $\mu$ (International Tables for X-ray Crystallography, 1959), and therefore it seems useful to extend the treatment of the absorption correction to the general case.

In what follows we will only describe a procedure for calculating the direction cosines of the incident and diffracted beams in a reference system attached to the crystal; from this point on the calculation of the transmission factor for each reflection can be performed as suggested, for example, by Busing \& Levy (1957) or by Wuensch \& Prewitt (1965).

It has been shown (Santoro \& Zocchi, 1966) that, for any Weissenberg method, a reciprocal lattice point is in reflecting position when:

$$
\begin{array}{r}
z s\left(n-2 v_{0}\right)+2\left[1-s^{2}\left(\frac{n}{2}-v_{0}\right)^{2}\right]^{\frac{1}{2}}(y \cos \omega-x \sin \omega) \\
+d^{* 2}=0
\end{array}
$$

$$
+d^{* 2}=0, \quad \text { (1) }
$$

where $\omega$ is the rotation angle, $v_{0}$ is the index of the reciprocal layer under examination, $s$ is the period on the rotation axis, $x, y, z$ are the coordinates of the point for $\mu=\omega=0^{\circ}$ with respect to a Cartesian system $X, Y, Z$, attached to the laboratory, and defined in the given reference,
$d^{* 2}=x^{2}+y^{2}+z^{2}$, and $n=(2 \sin v) / s$ (International Tables).
From equation (1) we obtain:

$$
\begin{equation*}
\tan \frac{\cdots}{2}=\frac{x \pm\left[x^{2}-(A+y)(A-y)\right]^{\frac{1}{2}}}{A-y} \tag{2}
\end{equation*}
$$


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